

CHEM/ATMS 458

Problem Set #2 Models and Transport

Due: Wednesday October 27, 2010

Problems from Text done with pen and paper

3.1, 3.2, 3.6, 6.5

Additional problem: Global CH₄

CH₄ is one of the most important anthropogenic greenhouse gases but it has significant biogenic sources ~ 50% of the total. Its present day mixing ratio in the northern hemisphere (NH) is ~ 1800 ppb while in the southern hemisphere (SH) it is 1600 ppb. We will use this observed gradient to determine the difference between sources in the NH and the SH. We take as the timescale for mixing air between the two hemispheres to be 1 year and the CH₄ lifetime by oxidation in the atmosphere is ~ 9 years.

- a) Given the above methane lifetime, is it valid to assume CH₄ is well mixed throughout each hemisphere?
- b) Using a two-box model for the NH and SH, write the mass balance equations for CH₄ in these two boxes.
- c) Assume that the gradient between the hemispheres is constant, i.e. $d(m_N - m_S)/dt = 0$, where m_N is the amount of methane in the NH and m_S is the amount in the SH. Calculate ΔE , the *difference* between NH and SH methane emission rates in units of moles of methane per year.
- d) If the total global emission rate of CH₄ $E_{Tot} = E_N + E_S \sim 5 \times 10^{13}$ moles/yr, what do you conclude about the distribution of methane sources between the two hemispheres? Does your calculation support a large terrestrial (land-based) component to the methane source? Why or why not?